SYNTHESES OF SOME DI-, TETRA-, AND HEXADEUTERIO-OCTADECENOATES

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#### SUMMARY

(E&Z)-10-Octadecenoic-13,14- $\underline{d}_2$ , -14,15- $\underline{d}_2$ , and -14,14,15,15- $\underline{d}_4$  acids were prepared by Wittig synthesis. (E&Z)-11-Octadecenoic-14,14,15,15,17,18- $\underline{d}_6$  acids were prepared by Wittig synthesis and by synthesis and reduction of 11octadecynoic-14,14,15,15,17,18- $\underline{d}_6$  acid. Overall yield in 11-octadecenoate- $\underline{d}_6$  synthesis was better by the Wittig procedure (25% vs. 15%). The Wittig-produced 11-octadecenoates had higher hexadeuterio contents than the acetylenic sequence products (85% vs. 70%) but the average isotopic purity was similar (5.9 D/mole). The Wittig method appears better for deuterium-labelled octadecenoate synthesis.

KEY WORDS: Methyl (E&Z)-10-octadecenoates-13,14- $\underline{d}_2$ , methyl (E&Z)-10-octadecenoates-14,15- $\underline{d}_2$ , methyl (E&Z)-10-octadecenoates-14,14,15,15- $\underline{d}_4$ , 11-octadecynoic acid-14,14,15,15,17,18- $\underline{d}_6$ , methyl (E&Z)-11-octadecenoates-14,14,15,15,17,18- $\underline{d}_6$ .

### INTRODUCTION

A series of deuterated octadecenoic acids have been synthesized for studies of the distribution of isomeric octadecenoic acids in human blood fractions (1-5). With the synthesis reported here, the preparation of the (E&Z)-isomers having double bonds in positions 8-13 and with 2, 4, or 6 deuterium atoms has been completed.

\*The mention of firm names or trade products does not imply that they are endorsed or recommended by the U.S. Department of Agriculture over other firms or similar products not mentioned.

## RESULTS AND DISCUSSION

Tetradeuterated methyl (E&Z)-10-octadecenoates were prepared from 1-chloro-... 4-octyne (Figure 1, left column). The chlorooctyne was saturated with deuterium gas employing tris(triphenylphosphine)chlororhodium. The resulting 1-... chlorooctane-4,4,5,5- $\underline{d}_4$  was converted to the iodide and then to the corresponding triphenylphosphonium iodide. The phosphonium iodide by Wittig reaction with methyl 10-oxodecanoate gave methyl (E&Z)-10-octadecenoates of which 93% was the (Z)-isomer. The (E)-isomer was additionally obtained by stereoisomerization of the (Z)-isomer.

Methyl (E&Z)-10-octadecenoates-14,15- $\frac{d}{-2}$  were prepared from 1-chloro-4octyne by partial reduction to 1-chloro-4-octene and then application of a similar reaction sequence to that used in preparing the tetradeuterated isomers (Figure 1, right column).

## Figure 1

Preparation of Methyl (E&Z)-10-Octadecenoates-14,15-d2

and -14,14,15,15-d4

$$C1(CH_{2})_{3}C=C(CH_{2})_{3}H$$

$$H_{2}, Lindlar$$

$$C1(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 99\%$$

$$C1(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 99\%$$

$$C1(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 99\%$$

$$C1(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 99\%$$

$$C1(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 86\%$$

$$\int 0HC(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{3}H = 83\%$$

$$H(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

$$H(CH_{2})_{3}(CDD_{2})_{2}(CH_{2})_{2}CH=CH(CH_{2})_{8}CO_{2}CH_{3}$$

Methyl (E&Z)-10-octadecenoates-13,14- $\underline{d}_2$  were prepared from the tetrahydropyranyl (THP) ether of 3-octyn-1-ol by Lindlar-catalyzed partial reduction followed by deuteration with RhCl(Ph<sub>3</sub>P)<sub>3</sub> as catalyst. The resultant 2-(octyl-3,4- $\underline{d}_2$ ) tetrahydropyran was converted to the iodide and by subsequent Wittig synthesis to (E&Z)-10-octadecenoates 13,14- $\underline{d}_2$  in 42% yield.

# Figure 2

Preparation of (E&Z)-11-Octadecenoic-14,14,15,15,17,18-d<sub>6</sub> Acids  $HC = CCH_2 OTHP$  (THP =  $\bigcirc$  ) HC≡C(CH<sub>2</sub>)<sub>2</sub>OTHP 1) Et MgBr 2)  $H_2C=CH_2CH_2Br$ 1) EtMgBr ↓ 2) H<sub>2</sub>C=CH CH<sub>2</sub>Br H<sub>2</sub>C=CHCH<sub>2</sub>C≡CCH<sub>2</sub>OTHP  $H_2C=CHCH_2C=C(CH_2)_2OTHP$ 94% D<sub>2</sub> P<sub>2</sub> RhCl(PPh)<sub>3</sub> RhCl(PPh<sub>2</sub>) H(CHD<sub>2</sub>)CH<sub>2</sub>(CD<sub>2</sub>)<sub>2</sub>CH<sub>2</sub>OTHP  $H(CHD)_2CH_2(CD_2)_2(CH_2)_2OTHP$ (C6H7D60THP) 93% (C7H9D60THP) 93% ↓ кі, н<sub>3</sub>ро<sub>4</sub> ↓ NaI, H<sub>3</sub>PO<sub>4</sub>  $C_7 H_9 D_6 I 87\%$  $\downarrow Ph_3 P$ C6H7D61 87% LiC=C(CH<sub>2</sub>)<sub>9</sub>CO<sub>2</sub>Li  $\downarrow^{\text{LiC}=C(\text{CH}_{2}')_{9}^{\text{CO}}_{2}^$  $\downarrow^{\text{Reductions}} \qquad \qquad \downarrow^{\text{OHC}(CH_2)_9 \text{CO}_2 \text{CH}_3} \\ \text{C}_6\text{H}_7\text{D}_6\text{CH=CH}(CH_2)_9 \text{CO}_2\text{H}} \qquad \qquad \text{C}_6\text{H}_7\text{D}_6\text{CH=CH}(CH_2)_9 \text{CO}_2\text{CH}_3$ E or Z 43% (E&Z)-isomers

Hexadeuterated methyl (E&Z)-11-octadecenoates were prepared by the two sequences shown in Figure 2. An acetylenic synthesis was carried out as indicated in the left column of the figure. A new synthesis was developed to prepare 11-dodecynoic acid (7). The acetylenic acid was alkylated (8,9) with hexadeuterio-1-iodohexane to give 11-octadecynoic-14,14,15,15,17,18- $\underline{d}_6$  acid which was reduced with Li in ammonia to hexadeuterated (E)-11-octadecenoic acid,  $\underline{1}$ , and with Pd/BaSO<sub>4</sub> or with Lindlar's catalyst to the (Z) isomer,  $\underline{2}$ . An alternative synthesis of 11-octadecynoic- $\underline{d}_{-6}$  acid through hexadeuterio-1-octyne and its alkylation by 10-iododecanoic acid gave only 35% yield vs. 56% from alkylation of 11-dodecynoic acid. Attempts to alkylate 1-octyne- $\underline{d}_{-6}$  with the iodomagnesium salt of 10-iododecanoic acid gave low conversions.

Hexadeuterated (E&Z)-11-octadecenoates were also prepared by the Wittig sequence as indicated in the right-hand column of Figure 2. Alkylation of the bromomagnesium salt of the THP ether of 3-butyn-1-ol with allyl bromide gave the THP ether of 6-hepten-3-yn-1-ol which was deuterium-saturated and converted to 1-iodoheptane-3,3,4,4,6,7- $\frac{d}{-6}$ . Wittig reaction sequence with methyl 11-oxo-undecanoate gave a mixture of methyl esters of 1 and 2 (93% 2).

The hexadeuterio-contents (Table 1) of the methyl esters of  $\underline{1}$  and  $\underline{2}$  derived from the Wittig procedure were found to be higher than those of  $\underline{1}$  and methylated  $\underline{2}$  from reduction of 11-octadecynoic- $\underline{d}_6$  acid. Furthermore, reduction products  $\underline{1}$  and  $\underline{2}$  of the acetylenic acid, showed inexplicable scatter over that found in their common precursor, the THP ether of 1-hexanol- $\underline{d}_6$  (Table 1). The possibility of exchange is excluded by the relatively constant average deuterium per mole. Since 11-octadecynoic acid could not be analyzed for deuterium due to triple-bond migration, it was reduced with azodicarboxylic acid to a mixture of octadecanoic, octadecenoic, and octadecynoic acids. The octadecanoic- $\underline{d}_6$  acid to have scatter intermediate between hexyloxy- $\underline{d}_6$  THP and that of  $\underline{1}$  and  $\underline{2}$ . Thus, deuterium scatter appears to result during the alkylation and reduction steps. Based upon these results, it would appear that the better procedure for preparation of deuterium-labelled octadecenoates is the Wittig sequence.

TABLE	1

Deuterium distributions

Octadecenoate	Number of deuterium atoms <sup>a</sup>									
	0	1	2	3	4	5	6	7	8	Average
Methyl-10- <sup>b</sup>	2	1	3	4	85	0.3	1	1	0	3.9
Methyl-10- <sup>b</sup>	4	7	76	8	3	1	1	0	0	2.0
Methyl-10- <sup>b</sup>	6	2	89	1	1	0	0	0	0	1.9
Methyl-11- <sup>b</sup>	2	0	2	0	2	5	86	3	1	5.9
Methyl-(Z)-11 <sup>C</sup>	0	0	1	1	3	8	71	10	4	6.0
(E)-ll-acid <sup>C</sup>	0	0	2	1	4	8	71	9	3	5.9
2-(Hexyloxy) THP	0	0	0	0	1	6	88	5	0	5.9
Methyl octadecanoate <sup>C</sup>	1	1	1	1	2	8	75	10	1	5.9

<sup>a</sup> Table figures in percent, except under average.

<sup>b</sup> Wittig products (resp. 14,14,15,15- $\underline{d}_4$ , 14,15- $\underline{d}_2$ , 13,14- $\underline{d}_2$  and

14,14,15,15,17,18-<u>d</u>6.

<sup>c</sup> From 11-octadecynoic  $-\underline{d}_6$  and reduction.

## EXPERIMENTAL

The structures of synthesized compounds were confirmed by  ${}^{1}$ H and  ${}^{13}$ C NMR spectra and/or IR. Melting points are uncorrected. Preparative argentation separations were carried out on a 6 ft x 2 inch column packed with fully-silvered XN1010 ion exchange resin.

 $\frac{1-\text{Chlorooctane-4,4,5,5-d}_{-4}}{\text{RhCl}(\text{Ph}_{3}\text{P})_{3} \text{ by previously used method (1) in 99\% yield.}}$ 

2-Iodooctane-4,4,5,5- $\underline{d}_4$ . Refluxing 1-chlorooctane- $\underline{d}_4$  with excess NaI in 2-butanone overnight gave a 93% yield.

<u>1-Octyl-4,4,5,5-d</u><sub>4</sub> triphenylphosphonium Iodide. Refluxing 1-iodooctane-<u>d</u><sub>4</sub> with excess  $Ph_3P$  in acetonitrile gave a 94% yield of phosphonium salt, m.p. 73-4°C.

Methyl (E&Z)-10-Octadecenoates-14,14,15,15- $\underline{d}_4$ . In a dry flask fitted with a  $N_2$ -inlet, stirrer, thermometer, and addition funnel protected by a drying tube were placed 120.9 g (0.24 mole) 1-octyltriphenylphosphonium iodide and 400 mL of dry DMF. The mixture was stirred under  $\mathrm{N}_{\mathrm{p}}$  until the solid dissolved and then 12.9 g (0.239 mole) of freshly prepared NaOCH, was added. The orange slurry was stirred ½ hr and then 45.2 g (0.239 mole) methyl 10oxodecanoate was added over 1 hr. The mixture was stirred 3/4 hr and then poured into 1 L pH 2 water. The product was taken up by extraction with 5, 100-mL portions of PE (35-60°C). The PE extracts were washed with 1 L water and dried over Na $_2$ SO<sub>4</sub>. The oily residue from evaporation of the PE was distilled to give 38.7 g of oil (bp. 120-160°C/0.05 mm Hg). GC analysis indicated the oil to consist largely of methyl 10-octadecenoates (83%) and methyl 10-oxodecanoate (14.6%). The aldehyde-ester was largely removed by extraction of an ether solution of the esters with two 100-mL portions of saturated NaHSO, solution. This procedure gave 29.8 g of 95% pure methyl 10-undecenoates. Finally purification and separation of the E&Z-isomers was effected by argentation chromatography to yield 24.2 g of pure (Z)-isomer and 2.7 g of pure (E) isomer in 34 and 4% yields, respectively.

Additional amounts of the (E)-isomer were prepared by isomerization (10). Thus, from 24.2 g of (Z)-isomer there were obtained 16.2 g (67%) of pure (E)-isomer and 3.5 g (14.6%) of (Z)-isomer following isomerization and argentation chromatography.

 $\frac{1-\text{Chlorooctane-4,5-d}_2}{\text{marked}_2}$ . Deuteration of 88 g (0.6 mole) 1-chloro-4-octene with 7 g RhCl (Ph<sub>3</sub>P)<sub>3</sub> catalyst in 1 L benzene by a previous procedure (1) gave  $\ddot{}$ 78.0 g 1-chlorooctane-4,5-d<sub>2</sub> of 99.6% GC purity, 86.3% yield.

 $\frac{(1-0\text{ctyl}-4,5-\underline{d}_2) \text{ triphenylphosphonium Iodide. } 1-Chlorooctane-\underline{d}_2 (60.3 \text{ g}, 0.4 \text{ mole}) \text{ was heated at reflux for 11 hr with 65 g (0.43 mole) NaI in 350 mL CH<sub>3</sub>CN. Then 113 g (0.43 mole) Ph<sub>3</sub>P was added and the mixture was filtered and the filtrate triturated with 1 L and two 500-mL portions of ether. Each portion of ether was decanted. The crystals which formed were filtered and$ 

stirred with three 500-mL portions of ether. The yield was 188.5 g (93.4%)
m.p. 74-75°C.

Methyl (E&Z)-10-Octadecenoates-14,15- $\underline{d}_2$ . Wittig synthesis between 141 g (0.28 mole) 1-octyl-4,5- $\underline{d}_2$ -triphenylphosphonium iodide and 60 g (0.3 mole) methyl 10-oxo-decanoate as described above for preparation of 10-octadecenoates-14,14,15,15- $\underline{d}_4$  gave 31.9 g (60%) of (E&Z)-10-octadecenoates- $\underline{d}_2$ . The isomers were separated by argentation chromatography to give 18.4 pure (Z)-isomer and 1.9 g (E)-isomer. Isomerization (10) of 18.5 g of a mixture of the (E&Z)isomers followed by argentation chromatography gave 12.3 g of the (E)-isomer.

<u>2-(3-Octenyloxy)tetrahydropyran</u>. Hydrogenation of 2(3-octynyloxy) tetrahydropyran over Lindlar's catalyst gave a quantitative yield of 2-(3octenyloxy)tetrahydropyran (pure by GC).

 $\frac{2-(0ctyl-3,4-\underline{d}_2)tetrahydropyran.}{98\%}$  yield of 2-(octyl-3,4-\underline{d}\_2) tetrahydropyran.

 $\frac{1-\text{Iodooctane-3,4-d}_2}{\text{gave 1-iodooctane-3,4-d}_2}$ . Treatment of the previous product with  $\text{H}_3\text{PO}_4$ -KI (12) gave 1-iodooctane-3,4-d\_2 in 73% yield.

<u>1-Octyl-3,4-d</u> triphenylphosphonium iodide. The previous product refluxed in acetonitrile overnight with 10% excess  $Ph_3P$  gave a 99% yield of the phosphonium salt (m.p. 75-77°C) following purification by trituration in ether.

Methyl (E&Z)-10-Octadecenoates. Wittig synthesis between 1-octyl-3,4- $\frac{d}{2}$ triphenylphosphonium iodide and methyl 10-oxodecanoate as for 10-octadecenoate- $\frac{d}{4}$  above gave a 42% yield of (Z&E)-10-octadecenoates-13,14- $\frac{d}{2}$  (90% Z) which were separated by argentation chromatography.

<u>2-(5-Hexene-2-ynyloxy)tetrahydropyran</u>. This compound was prepared by a published method (11) in 80% yield.

 $\frac{2-(\text{Hexyloxy-2,2,3,3,5,6-d_6})\text{tetrahydropyran.}}{\text{hexen-2-ynyloxy}\text{tetrahydropyran was achieved by use of RhCl (Ph_3P)_3 (1) after 20 hr in 93% yield.}$ 

 $\underbrace{1-\text{Iodohexane-2,2,3,3,5,6-d_6}}_{\text{with } \text{H}_3\text{PO}_4-\text{NaI}} (12) \text{ gave } 87\% \text{ of the corresponding iodide.}}$ 

<u>10-Hydroxydecanoic Acid</u>. A flask fitted with gas-dispersion tube, mechanical stirrer, thermometer, and reflux condenser and cooled with a Dry Ice bath was charged with a solution of 198.3 g (1.0 mole) methyl 10-undecenoate in 1.5 L methanol. At 10°C 3% ozone in oxygen was bubbled through at 3 L/min. After 4 hr, the ozone flow was stopped and the solution was sparged for 15 min with  $N_2$ . Sodium borohydride (45 g) was added in portions at 5-10°C. The solution was allowed to stand for 13 hr and then heated at reflux with 200 mL 5N aq. KOH. About 1 L methanol was evaporated and the residue was taken up in 1.5 L  $H_2^0$  and acidified with 5N  $H_2^{SO_4}$ . The product was extracted with 6, 150 mL portions of ether, washed with saline, and dried over  $Na_2^{SO_4}$ . The ether solution was filtered, concentrated to 400 mL and diluted with 600 mL PE. The separated crystals (211.8 g) had m.p. 70-2°C. Recrystallization from PE-ether gave 155.3 g (77%) colorless solid, m.p. 71-3°C (reported 75-6°C) (13). GC analysis of a methylated sample showed a purity of 93%, sufficient for further synthesis.

<u>10-Iododecanoic Acid</u>. Treatment of 10-hydroxydecanoic acid with  $H_3PO_4$ -KI (12) gave 95% 10-iododecanoic acid, m.p. 46-9°C (m.p. 49-50°C) (14).

<u>ll-Dodecynoic Acid</u>. Treatment of ll-iododecanoic acid with lithium acetylide ethylenediamine (6) gave 86% of ll-dodecynoic acid, m.p. 44-45°C (reported 44-46°C) (15).

11-Octadecynoic-14,14,15,15,17,18- $\underline{d}_6$  Acid. Alkylation (8,9) of 11dodecynoic acid with 1-iodohexane-2,2,3,3,5,6- $\underline{d}_6$  gave the  $\underline{d}_6$ -acid in 46% yield, crystals from PE-ether, m.p. 45-46°C (reported 45.5-46.5°C for undeuterated acid) (17).

(E)-11-Octadecenoic-14,14,15,15,17,18-d<sub>6</sub> Acid (<u>1</u>). This acid was obtained in 77% yield by reduction with Li in liquid ammonia according to Barve & Gunstone (17), m.p. 43-43.5°C identical to that of the undeuterated isomer.

 $(2)-11-0ctadecenoic-14,14,15,15,17,18-\underline{d_6} \text{ Acid } (\underline{2}). A \text{ sample } (0.5 \text{ g}) \text{ of } 11-octadecynoic-\underline{d_6} \text{ acid was reduced in 10 mL benzene over } 0.1 \text{ g. Lindlar's }$ 

of the PE gave 2.0 g which was partially separated by argentation chromatography. The methyl octadecanoate fraction was analyzed by MS.

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catalyst and 0.1 g dry quinoline. The hydrogenation was monitored with an automatic manometer and gave theoretical uptake in  $\frac{1}{2}$  hr. GC analysis of a methylated sample showed 2.2% acetylenic acid remaining and 1.5% of  $\frac{1}{2}$ .

A similar sample (0.5 g) of 11-octadecynoic- $\underline{d}_6$  acid was reduced in 10 ml pyridine with 0.1 g 5% Pd/BaSO<sub>4</sub>. Reduction was completed in  $\frac{1}{2}$  hr. Analysis by argentation chromatography and GC indicated ca. 8.1% stearate and 8.5%  $\underline{1}$  were present. A sample was purified by argentation chromatography for MS.

<u>2-(6-Hepten-3-ynyloxy)tetrahydropyran</u>. Treatment of 2(3-butynyloxy) tetrahydropyran with ethylmagnesium bromide and then allyl bromide (11) gave an 84% yield, purity 98% (GC).

2-(1-Heptyloxy-3,3,4,4,6,7-d\_6)tetrahydropyran. Deuteration of hepten-3ynyloxytetrahydropyran by previous method (1) gave a 93% yield.

 $\frac{1-\text{Iodoheptane-3,3,4,4,6,7-d}_{6}}{\text{with } \text{H}_{3}\text{PO}_{6}-\text{NaI}}$  (3) gave 87% of the corresponding iodide.

 $\frac{1-\text{Heptyl}-3,3,4,4,6,7-\underline{d}_6}{\text{triphenylphosphonium Iodide.}}$  The phosphonium salt was obtained in 94% yield (2), m.p. 129-132°C, by refluxing the previous product with 10% excess Ph<sub>3</sub>P as for octyl-\underline{d}\_4 phosphonium salt above.

Methyl 11-Oxoundecanoate. Methyl 11-hydroxyundecanoate was oxidized (18) to give the aldehyde-ester in 80 to 96% yields.

Methyl (E&Z)-11-Octadecenoates-14,14,15,15,17,18- $\underline{d}_6$ . Wittig synthesis between methyl 11-oxoundecanoate and heptyl- $\underline{d}_6$  phosphonium salt gave 43% of a mixture of (E&Z)-isomers (93% Z). The octadecenoates were separated by argentation chromatography. Methyl (E)-11-octadecenoate- $\underline{d}_6$  was additionally prepared by isomerization of the (Z)-isomer (10).

Methyl Octadecanoate-14,14,15,15,17,18- $\underline{d}_6$ . A flask fitted with magnetic stirrer, N<sub>2</sub>-inlet, drying tube and addition funnel was charged with 2.0 g (10.3 mmole) potassium diazodicarboxylate. A solution of 0.19 g (0.65 mmole) methyl octadecynoate- $\underline{d}_6$  in 30 ml dry pyridine was added. The resultant slurry was vigorously stirred under N<sub>2</sub> for 2 hr while 2 ml (35 mmole) acetic acid was added. The mixture was decomposed in water and extracted with PE. Evaporation